

# $3\alpha$ -(4-Substituted Phenyl)nortropane- $2\beta$ -carboxylic Acid Methyl Esters Show Selective Binding at the Norepinephrine Transporter

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**Abstract**—A series of  $3\alpha$ -(4-substituted)nortropane-2β-carboxylic acid methyl esters was synthesized and evaluated for the ability to inhibit radioligand binding at the dopamine, serotonin, and norepinephrine transporters.  $3\alpha$ -(4-Methylphenyl)nortropane-2β-carboxylic acid methyl ester (**4c**) was found to be selective and highly potent for the norepinephrine transporter (NET) relative to the dopamine and serotonin transporters. © 2000 Elsevier Science Ltd. All rights reserved.

The pharmacology of cocaine (1) is believed to center around its interaction with the dopamine, serotonin, and norepinephrine transporters (DAT, 5-HTT, and NET, respectively). As part of a program to study the biochemical mechanism of action of cocaine, we have conducted a structure–activity relationship (SAR) study to investigate the monoamine transporter binding properties of the 3-phenyltropane class of compounds.<sup>1</sup> These studies have led to the discovery of analogues selective for the DAT such as 3β-(4-chlorophenyl)-2β-(3-phenylisoxazol-5'-yl)tropane (2, RTI-177)<sup>2,3</sup> as well as analogues selective for the 5-HTT such as 3β-(4ethyl-3-iodophenyl)nortropane-2β-carboxylic acid methyl ester (3, RTI-353, EINT) (Chart 1).<sup>4,5</sup> As a continuation of these SAR studies, we now report the synthesis of the 3α-(4-substituted phenyl)nortropane-2β-carboxylic acid methyl esters (4a-c), some of which possess greater affinity at the NET than that at the DAT and 5-HTT.

## Chemistry

The  $3\alpha$ -(4'-fluoro, -chloro, and -methylphenyl)nortropane-2 $\beta$ -carboxylic acid methyl esters (4a–c, respectively) were synthesized by refluxing the known  $3\alpha$ -(4'-substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl esters (5a–c)<sup>6</sup> with  $\alpha$ -chloroethyl chloroformate (ACE-Cl) in dichloroethane under a nitrogen atmosphere to give an (α-chloroethyl)urethane, which was not isolated but converted directly to the *N*-nor analogues by solvolysis with methanol (Scheme 1).<sup>7</sup> Concentration of the reaction mixture followed by the addition of ethyl ether provided **4a**–**c** as their hydrochloride salts. Each compound gave satisfactory elemental analyses for carbon, hydrogen, and nitrogen, and the <sup>1</sup>H NMR spectra were in

2 (RTI-177)

Chart 1.

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CH<sub>3</sub>, 
$$CO_2CH_3$$

$$X$$
ACE-CI, DCE, reflux;
$$CH_3OH, reflux$$

$$X$$

$$ACE-CI, DCE, reflux;$$

$$CH_3OH, reflux$$

$$X$$

$$Aa, X = F$$

$$b, X = CI$$

$$c, X = CH_3$$

$$CO_2CH_3$$

$$X$$

CH<sub>3</sub> N ACE-Cl, DCE, reflux; CO<sub>2</sub>CH<sub>3</sub> ACE-Cl, DCE, reflux; CO<sub>2</sub>CH<sub>3</sub> 
$$X$$

6a,  $X = F$ 
b,  $X = Cl$ 
c,  $X = CH_3$ 

ACE-Cl, DCE, reflux; CO<sub>2</sub>CH<sub>3</sub>

7a,  $X = F$ 
b,  $X = Cl$ 
c,  $X = CH_3$ 

Scheme 1.

agreement with the assigned structures. The  $2\beta$ ,  $3\beta$ -phenyltropanes, **6a–c**, and  $2\beta$ ,  $3\beta$ -phenylnortropanes, **7a–c**, were prepared as previously described. Depiction of the  $2\beta$ ,  $3\alpha$  isomers as boats and  $2\beta$ ,  $3\beta$  isomers as chairs is intentional. Previous NMR analysis suggests that the ring of the  $2\beta$ ,  $3\beta$  compounds inverts when the center is epimerized to the  $2\beta$ ,  $3\alpha$  isomers.

## **Biological Studies**

The in vitro binding affinities of the compounds at the DAT, 5-HTT, and NET were determined via competitive binding assays using the previously reported procedures. The radioligands were 0.5 nM [ $^3$ H]WIN 35,428 for the DAT, 0.2 nM [ $^3$ H]paroxetine for the 5-HTT, and 0.5 nM [ $^3$ H]nisoxetine for the NET. The binding data for the 2 $\beta$ ,3 $\alpha$ -phenylnortropane analogues, 4a-c, along with previously reported data of the 2 $\beta$ ,3 $\beta$ -phenyltropane (6a-c), 2 $\beta$ ,3 $\beta$ -phenylnortropane (7a-c), and 2 $\beta$ ,3 $\alpha$ -phenyltropane (5a-c) analogues as well as cocaine (1) for comparison are given in Table 1.

#### Results and Discussion

A SAR study of the 3β-(4'-substituted phenyl)nortropane-2β-carboxylic acid methyl esters revealed that removing the *N*-methyl group from the tropane analogue resulted in increased binding at NET and 5-HTT with little change in binding at the DAT.<sup>4,7</sup> This can be seen by comparing the affinity of the 3-phenyltropane analogues, **6a**-**c**, to the nortropane analogues, **7a**-**c** (Table 1), respectively. In all three cases, binding at the DAT remained relatively constant while binding to the

 $\textbf{Table 1.} \quad \text{Comparison of transporter binding properties of 3-phenyltropane and 3-phenylnortropane } 2\beta\text{-carboxylic acid methyl ester analogues}$ 

R. N. 
$$CO_2CH_3$$
 $2\beta,3\alpha$ -isomer

RTI compound <sup>b</sup>	Isomer				$IC_{50}$ nM $(K_i$ nM) <sup>a</sup>		
	2	3	R	X	NE [ <sup>3</sup> H]nisoxetine	DA [ <sup>3</sup> H]WIN 35,428	5-HT [ <sup>3</sup> H]paroxetine
WIN 35, 428 (6a) <sup>c</sup>	β	β	CH <sub>3</sub>	F	835±45 (503±27)	15.7±1.4	760±47 (69±4)
142 ( <b>7a</b> ) <sup>c</sup>	β	β	Н	F	$18.80\pm0.68$ (11.3±0.41)	4.4±0.2	$68.6\pm2.0$ $(6.24\pm0.18)$
286 ( <b>5a</b> ) <sup>d</sup>	β	α	$CH_3$	F	$1200\pm91$ (741±54.8)	$21.0 \pm 0.50$	5060±485 (460±44)
367 ( <b>4a</b> )	β	α	Н	F	$9.8\pm0.7$ $(5.9\pm0.40)$	32.6±2.6	$92.4\pm7.7$ $(8.40\pm0.70)$
31 <b>(6b)</b> <sup>c</sup>	β	β	$CH_3$	Cl	$37\pm2.1$ (22.0±1.3)	1.12±0.10	$45.0\pm1.3$ $(4.00\pm0.12)$
110 ( <b>7b</b> ) <sup>c</sup>	β	β	Н	Cl	$5.45\pm0.21$ (3.28±0.13)	$0.62 {\pm} 0.09$	$4.13\pm0.62$ (0.38±0.06)
355 ( <b>5b</b> ) <sup>d</sup>	β	α	$CH_3$	Cl	$60\pm 2.40$ (36.0±1.5)	2.4±0.2	998±120 (91±11)
389 ( <b>4b</b> )	β	α	Н	Cl	$5.14\pm1.08$ (3.1±0.60)	3.1±1.0	$53\pm 3$ (4.80±0.26)
32 ( <b>6c</b> ) <sup>d</sup>	β	β	$CH_3$	$CH_3$	$60.0\pm0.50$ (36.0±0.30)	$1.70 \pm 0.30$	$240\pm27$ (22.0±2.5)
404 ( <b>7c</b> )	β	β	Н	$CH_3$	$7.20\pm0.45$ (4.40±0.27)	$0.84{\pm}0.09$	135±28 (12±3)
356 ( <b>5c</b> ) <sup>d</sup>	β	α	$CH_3$	$CH_3$	$270\pm24$ (160±14)	$10.2 {\pm} 0.8$	$4250\pm422$ (390±38)
362 ( <b>4c</b> )	β	α	Н	$CH_3$	$9.0\pm0.3$ (5.20±0.18)	33.6±4.1	500±30 (46±3)
Cocaine (1)	_	_	_	_	$3300\pm290$ (1900±170)	89.1±4.8	$1050\pm 89$ (95±8)

<sup>&</sup>lt;sup>a</sup>The numbers under the IC<sub>50</sub> value in parentheses are the  $K_{\rm I}$  values.

<sup>&</sup>lt;sup>b</sup>Compounds **4a-c**, **5a** and **6a-c** and **5b-c** were assayed as their hydrochloride, tartrate, and tosylate salts, respectively. Compounds **7a-c** were assayed as free bases.

<sup>&</sup>lt;sup>c</sup>The IC<sub>50</sub> values are from ref 7.

<sup>&</sup>lt;sup>d</sup>The IC<sub>50</sub> values are from ref 6.

NET increased 44-, 6.8-, and 8.3-fold, and binding to the 5-HTT increased 11-, 11-, and 2-fold for the 4'-fluoro, 4'-chloro, and 4'-methyl analogues, respectively. In a separate SAR study we reported that  $3\alpha$ -(4'-substituted phenyl)tropane- $2\beta$ -carboxylic acid methyl esters showed decreased binding at all three transporters relative to the corresponding  $2\beta$ ,  $3\beta$ -isomer; however, binding at the NET was sometimes affected less than the other two transporters. This trend is evidenced when comparing compounds 6a-c to 5a-c (Table 1). The data shows that epimerization at the 3-position caused binding to the NET to decrease only 1.4- to 4.5-fold, but binding to the DAT and 5-HTT became worse.

Since previous binding studies have shown that trends of the WIN 35,065-2 analogues are often additive, combining these two trends would suggest that  $3\alpha$ -(substituted phenyl)nortropane-2β-carboxylic acid methyl esters might be more potent and selective for the NET than at the DAT and 5-HTT. As expected,  $3\alpha$ -(4'-fluoro, -chloro, and -methylphenyl)nortropane-2β-carboxylic acid methyl esters, 4a-c, were found to have the highest potency for NET. A comparison of the affinity at the NET of the  $2\beta$ ,  $3\alpha$ -phenylnortropane analogues to that of the  $2\beta$ ,  $3\beta$ phenylnortropanes reveals that they have approximately the same affinities. For example,  $3\beta$ -(4'-chlorophenyl) nortropane-2β-carboxylic acid methyl ester 7b was found to bind with a  $K_i$  value of 3.3 nM, while its  $2\beta$ ,  $3\alpha$ nortropane isomer 4b possesses a  $K_i$  value of 3.1 nM. The NET  $K_i$  values for the 4'-methyl analogues, 7c and 4c, were 4.4 and 5.2 nM, respectively. The fluoro analogue 4a was the only analogue to show an increase in potency, roughly 2-fold over its epimer. The 4-methylphenyl analogue 4c was found to bind to the NET 7 and 9 times better than at the DAT and the 5-HTT, respectively. To our knowledge, **4c** is the first 3-phenyltropane analogue to show selectivity for the NET and, thus, represents a lead structure for the development of even more NET-selective analogues. The 4-fluorophenyl analogue 4a shows approximately equal affinity for the NET and 5-HTT with 6-fold selectivity relative to the DAT. The 4-chlorophenyl analogue 4b showed about equal affinity for all three transporters.

In summary, we have compared the monoamine transporter binding properties of the  $2\beta$ ,  $3\beta$ - and  $2\beta$ ,  $3\alpha$ -isomers of 3-(4-substituted phenyl)tropane-2-carboxylic

acid methyl ester to the corresponding 3-(4-substituted phenyl)nortropane-2-carboxlic acid methyl esters and have shown that the  $2\beta$ ,3 $\alpha$ -nortropane analogues, **4a**–**c**, possess the greatest selectivity for the NET.  $3\alpha$ -(4-Methylphenyl)nortropane-2 $\beta$ -carboxylic acid methyl ester (**4c**) is the first 3-phenyltropane analogue to show high potency and selectivity at the NET relative to the DAT and 5-HTT.

Additional analogues are currently under investigation to exploit these trends further. The discovery of NET-selective compounds also completes a set of WIN 35,065-2 compounds selective for each transporter affected by cocaine.

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